# Unusual Heme Structure in Cytochrome $aa_3$ from Sulfolobus acidocaldarius: A Resonance Raman Investigation<sup>†</sup>

George E. Heibel,<sup>‡</sup> Pavel Anzenbacher,<sup>§,||</sup> Peter Hildebrandt,<sup>\*,‡</sup> and Günter Schäfer<sup>§</sup>

Max-Planck-Institut für Strahlenchemie, Stifstrasse 34-36, D-45470 Mülheim a.d. Ruhr, Federal Republic of Germany, and Institut für Biochemie, Medizinische Universität Lübeck, Ratzeburger Allee 160, D-23562 Lübeck, Federal Republic of Germany

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ABSTRACT: Well-resolved Soret-excited resonance Raman spectra have been obtained in the heme marker band region (1440–1700 cm<sup>-1</sup>) for the fully oxidized and fully reduced forms of the terminal oxidase  $aa_3$  from Sulfolobus acidocaldarius. The results demonstrate that in both heme groups the structural properties depart from those of other  $aa_3$  oxidases. In the fully oxidized form, the formyl stretching vibration of heme a is observed at 1656 cm<sup>-1</sup>, approximately 7 cm<sup>-1</sup> higher in frequency than found for beef heart cytochrome c oxidase. The frequency of this vibration shows a slight upshift to 1657 cm<sup>-1</sup> in the fully reduced form. Thus, the formyl group of the heme a seems not to be involved in a significant hydrogen bond, in sharp contrast to both beef heart cytochrome c oxidase and other quinol-oxidizing enzymes. For heme  $a_3$ , two conformers were detected in the fully oxidized state. The observation of two  $\nu_3$  modes at 1482 and 1490 cm<sup>-1</sup> indicate the coexistence of both the normal six-coordinated high spin and a new five-coordinated high spin configuration. Both heme  $a_3$  species exhibit different formyl stretching vibrations at 1673 and 1666 cm<sup>-1</sup>, respectively. In the fully reduced enzyme, the identification of two heme  $a_3$  conformers is not unambiguous.

Sulfolobus acidocaldarius is a member of a group of aerobic archaebacteria living at high temperature in low pH media (Brock et al., 1972). Its ability to survive under these conditions makes it an interesting organism from the standpoint of cellular energetics. Previous studies have determined that it creates a large proton gradient across the plasma membrane, and this feature is coupled to proton pumps, driven by respiratory activity (Moll & Schäfer, 1988). These aspects of energy transduction are of current interest, and more experimental data on the structural and mechanistic details of the various proton pumps are required for a proper understanding of their functioning (Babcock & Wikström, 1992; Malmström, 1990; Wikström et al., 1981).

The terminal oxidase from S. acidocaldarius (SOX)<sup>1</sup> was first isolated as a single-subunit cytochrome  $aa_3$  with an apparent molecular mass of 38–40 kDa (Anemüller & Schäfer, 1989). This polypeptide was later identified as the product of the SOX-B gene in an operon coding for a larger electron transport complex also containing the gene (SOX-C) for an additional heme protein (Lübben et al., 1992). According to its amino acid sequence, the  $aa_3$  polypeptide has a true molecular mass of 57.9 kDa and, like well-characterized terminal oxidases, exhibits 12 putative membrane-spanning helices. The single polypeptide oxidase shows similarity with

eukaryotic cytochrome oxidases in terms of electronic absorption spectra (Anemüller & Schäfer, 1990). The redox centers have been proposed to consist of heme a and heme a3 as well as one copper ion, identified as Cu<sub>B</sub> of the binuclear center. The hemes in SOX show midpoint potentials similar to those of the corresponding hemes in beef heart cytochrome c oxidase (BOX). However, the former protein oxidizes reduced caldariella quinone, a thiophenobenzoquinone analog of ubiquinone found in the membranes of Sulfolobus (DeRosa et al., 1977), but it shows no activity with ferrocytochrome c. Known quinol oxidases differ from aa3-type oxidases most prominently in that the former lack Cu<sub>A</sub> and appear to oxidize quinols directly with heme a (Gennis, 1991; Lauraeus et al., 1991; Puustinen et al., 1991; Anemüller et al., 1992).

Resonance Raman (RR) spectroscopy has achieved considerable success in the elucidation of many important aspects of heme protein structure and function (Spiro & Li, 1988). The responsive behavior of certain bands in the spectral region between 1450 and 1650 cm<sup>-1</sup> (the marker band region) to changes in heme oxidation, spin, and coordination state has been shown to be extremely reliable, and its understanding has reached a highly predictive level (Kitagawa & Ozaki, 1987; Spiro & Li, 1988). Applications of RR spectroscopy to the study of cytochrome oxidases have revealed that heme a is six-coordinate, low spin (6cLS) in both oxidation states and that heme  $a_3$  is six-coordinate, high spin (6cHS) in the oxidized state and five-coordinate, high spin (5cHS) in the reduced state (Callahan & Babcock, 1981; Choi et al., 1983; Ogura et al., 1985; Ching et al., 1985; Babcock, 1988). This coordination change is related to the function of heme  $a_3$  in BOX, which is to bind  $O_2$  and allow a controlled reduction.

The present study extends our previous RR observations (Hildebrandt et al., 1991), employing higher resolution and an improved sample preparation which allows more detailed examination of the heme organization in this novel oxidase. We investigate the fully oxidized (SOX<sub>ox</sub>) and fully reduced (SOX<sub>red</sub>) forms of the terminal oxidase in the marker band

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<sup>\*</sup> To whom correspondence should be addressed.

<sup>\*</sup> Max-Planck-Institut für Strahlenchemie.

<sup>§</sup> Institut für Biochemie.

Present Address: Institute of Experimental Biopharmacy, Academy of Sciences of Czech Republic—Pro. Med. CS, Heyrovského 1207, 50002 Hradec Kralové, Czech Republic.

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Abbreviations: BOX, beef heart cytochrome c oxidase; SOX, Sulfolobus acidocaldarius cytochrome aa3; the subscripts "ox" and "red" denote the fully oxidized and fully reduced forms, respectively; RR, resonance Raman; 6cLS, six-coordinated low spin; 6cHS, six-coordinated high spin; 5cHS, five-coordinated high spin.

region (1440-1700 cm<sup>-1</sup>), which allows observation of the spin state marker bands as well as of substituent stretching vibrations.

#### MATERIALS AND METHODS

Biochemical Isolation and Sample Preparation. The terminal oxidase from S. acidocaldarius was isolated as described previously, with the following modifications. The chaotropic extraction of membranes was replaced by extraction with 30 mM sodium pyrophosphate in 50 mM malate buffer, pH 5.5; thereby the overall yield and spectral purity of the final product was largely enhanced. After hydroxyapatite chromatography, the aa<sub>3</sub>-containing fractions were concentrated on an Amicon PM-10 membrane and rechromatographed. Thereby, the overall yield and spectral purity of the product were significantly improved. The final preparation containing only the single-subunit form of the terminal oxidase from S. acidocaldarius was extensively dialyzed against a 1000-fold volume of 10 mM potassium phosphate, 10 mM EDTA, and 0.05% sarcosyl and reconcentrated thereafter. The optical absorption maxima were at 421 and 597 mm in the oxidized and at 439 and 601 nm in the reduced state (Anemüller & Schäfer, 1990). The metal content was determined by atomic absorption spectroscopy. The resulting preparations contained 1.6-2.1 mg of protein/mL with 1.9-2.3 mol of Fe/mol of aa<sub>3</sub>, and 0.45-0.5 mol of Cu/mol of Fe. Sulfolobus oxidase prepared as described has high quinol oxidase activity and does not contain CuA as also documented by the EPR spectra (Anemüller et al., 1992). Beef heart mitochondria were prepared according to the procedure described in Heibel et al. (1993). All samples were shock frozen in liquid nitrogen and stored at -60 °C. The thawed samples were diluted with buffer to an optical density of 1.0 at 413.1 nm. Reduction of the oxidase was achieved using a minimal amount of solid sodium dithionite.

Spectral Acquisition. Laser excitation at 413.1 nm was provided by a Kr<sup>+</sup> laser (Spectra Physics), operated at ca. 20 mW. The scattered light was collected at 90° from a rotating quartz cuvette and focused onto the entrance slit of a Spex 1404 double monochromator equipped with a photon counting system, including a cooled photomultiplier tube (RC 31034/ 76). The spectra were collected with a step size of 0.2 cm<sup>-1</sup> with an acquisition time of 0.5 s at each step. Spectral resolution was 3 cm<sup>-1</sup>. Photoreduction was not observed under our conditions. The intrinsic fluorescence of Sulfolobus aa3 which obscured much of the spectral information in previous studies was reduced by approximately 75% through the modified isolation procedure. All spectra were recorded at room temperature.

Data Handling. Spectra displayed are the sum of 8-12 scans, each scan requiring 20 min. The fluorescence background was subtracted using polynomial subtraction. For the summed spectra shown in this paper we have employed a filter which lowered the high-frequency noise level to the extent that subtraction of the filtered spectra from the original spectra resulted in evenly distributed noise with an amplitude of 0.5-1.0% of the average number of counts of the original spectra. Thus, the filtering procedure does not add or remove any spectral features from the spectra so treated.

Band Fitting. Spectra were analyzed with an interactive band fitting program based on a Marquardt algorithm using Lorentzian lines to fit the Raman bands. Other spectral lineshapes (Gaussian or Voigt functions) were not suitable for the spectral simulation, as previously noted in other studies

of the RR spectra of heme proteins (Hildebrandt & Stockburger, 1989; Hildebrandt et al., 1992; Heibel et al., 1993). The fit was achieved by progressively lowering the damping constants restraining the band parameters, until the difference between the measured and simulated spectra was within the noise level of the measured spectrum. This led to average standard deviations in SOX spectra of 0.8 cm<sup>-1</sup> in frequency and 3.0 cm<sup>-1</sup> in bandwidth, except for those bands in more crowded spectral regions (1470-1510 cm<sup>-1</sup> in SOX<sub>ox</sub>) for which the respective standard deviations in the band parameters for the latter bands were 1.5 and 6.0 cm<sup>-1</sup>. Standard deviations for BOX spectral parameters were smaller by a factor of approximately 2.5.

#### RESULTS AND DISCUSSION

The fully oxidized and fully reduced SOX were measured with 413-nm excitation in the marker band region (1450-1700 cm<sup>-1</sup>). Upon Soret band excitation, this region displays the heme a and heme  $a_3$  skeletal modes of  $A_{1g}$  ( $\nu_2$ ,  $\nu_3$ ,),  $B_{1g}$  $(\nu_{10}, \nu_{11})$ , and  $B_{2g}$   $(\nu_{28})$  symmetry [mode notation according to Abe et al. (1978)], which have been shown to be reliable indicators of core size, spin, oxidation, and ligation state of heme proteins in general and of cytochrome oxidases in particular (Babcock, 1988). The  $E_u$  ( $\nu_{37}$ ,  $\nu_{38}$ ) modes are also enhanced with this excitation wavelength and reflect the asymmetry of the heme substitution pattern and electronic distribution (Choi et al., 1983). In addition, this spectral region displays bands containing a dominant contribution of the porphyrin a peripheral substituent (formyl, vinyl) stretching vibrations. These bands are expected to reflect sensitively the immediate environment of the hemes, since they are most likely in direct contact with amino acid residues of the heme pocket. With 413-nm excitation, the resonance Raman (RR) spectra of aa<sub>3</sub>-type oxidases include contributions from both heme a and heme  $a_3$ , leading to a large number of overlapping bands in this region. This is reflected by the broad and asymmetric peaks in the spectrum of SOX<sub>ox</sub> shown in Figure 1A. Thus, it is helpful to compare this spectrum with that of BOX<sub>ox</sub> (Figure 1B), for which a detailed vibrational analysis has been carried out (Heibel et al., 1993).

The Fully Oxidized State. The RR spectrum of SOXox in Figure 1 is similar to that previously reported (Hildebrandt et al., 1991), although there are some differences which may be partly due to the improved isolation procedure in the present work. For example, the intensity of the 1589-cm<sup>-1</sup> peak is higher than in the previously reported spectrum. In addition, a higher structural differentiation of the lower intensity peaks below 1560 cm<sup>-1</sup> is also discernable in the present spectrum due to the significantly improved spectral resolution.

A comparison of SOX<sub>ox</sub> and BOX<sub>ox</sub> in Figure 1 shows a number of important differences. There are large intensity changes in all spectral regions, most obviously in the 1650and 1570-cm<sup>-1</sup> regions. The spectral region between 1460 and 1510 cm<sup>-1</sup> is clearly more crowded in the bacterial oxidase than in the beef heart enzyme. The highest frequency region, above 1645 cm<sup>-1</sup>, is very different for the bacterial enzyme. We have tried to ascribe these differences to the individual hemes in SOX<sub>ox</sub> by employing a band fitting analysis to deconvolute the bands in these spectra.

A comparison of the 1450-1560-cm<sup>-1</sup> region for BOX<sub>ox</sub> and  $SOX_{ox}$  is made in Figure 2. This region contains the  $\nu_3$ marker band which is relatively well isolated from interfering modes and is commonly used to determine the spin and coordination states of hemes in heme proteins. The RR bands

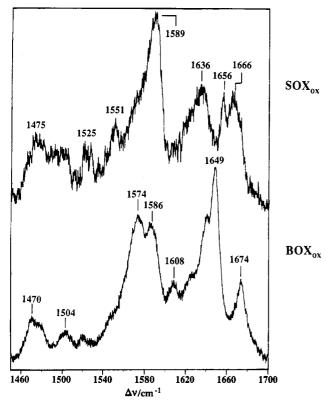


FIGURE 1: Resonance Raman spectra of (A) SOX<sub>ox</sub> and (B) BOX<sub>ox</sub> in the spectral region from 1450 to 1700 cm<sup>-1</sup> obtained with 413-nm excitation

for SOX<sub>ox</sub> are assigned in analogy to BOX<sub>ox</sub>, as far as possible (Heibel et al., 1993; Babcock, 1988; Table I). There are SOX bands corresponding to each BOX band in this spectral region, although the most remarkable difference between the two spectra is the observation of a peak located at ~1490 cm<sup>-1</sup> in SOX<sub>ox</sub> which has no counterpart in the RR spectrum of BOX<sub>ox</sub>. It is evident that this spectral region cannot be fitted by allowing the four existing BOX<sub>ox</sub> bands to broaden and fill the "valley" which is found at this position in BOX<sub>ox</sub>. The 1490-cm<sup>-1</sup> band occurs at a characteristic frequency for the  $\nu_3$  band of a 5cHS heme (Babcock, 1988). Consequently, we conclude that, in the Sulfolobus enzyme, one of the heme groups exists in a coordination equilibrium between a 5c and 6c state. In  $BOX_{ox}$  and other cytochrome c oxidases, heme a is ligated by two histidines which, by virtue of their strong ligand field character, yield a 6cLS configuration. In heme a<sub>3</sub>, a histidine and a weak field ligand, as yet unidentified, coordinate the central iron, leading to a 6cHS configuration in BOX. The evidence for a 5cHS population in SOX<sub>ox</sub> can be interpreted as an even more weakly binding heme  $a_3$  axial ligand.

These findings imply that in the entire spectral region we must take into account three different species: the 6cLS heme a and both the 6cHS and 5cHS configurations of heme  $a_3$ . This imposes serious difficulties on the vibrational analysis of the RR spectrum, in particular in the region between 1560 and 1630 cm<sup>-1</sup> which is expected to include up to 14 bands. In this region, a band fitting analysis cannot yield unambiguous results. Nevertheless, a qualitative examination of this region (Figure 1) also supports the conclusion of coexisting 5cHS and 6cHS states of heme  $a_3$ : the distinct peak at 1574 cm<sup>-1</sup> in BOX<sub>ox</sub>, which is due to the  $\nu_2$  peak of heme  $a_3$  (6cHS), is replaced in SOX<sub>ox</sub> by a much weaker but significantly broader shoulder of the 1589-cm<sup>-1</sup> peak ( $\nu_2$  of heme a) which can be readily understood in terms of lowering of the heme  $a_3$  6cHS

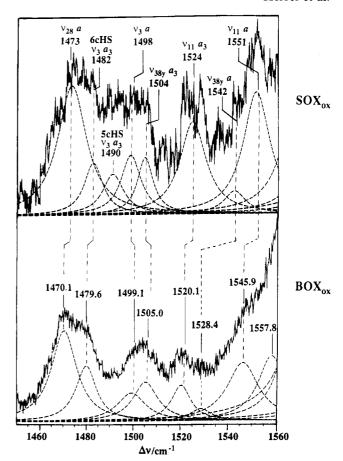


FIGURE 2: Resonance Raman spectra of (A) SOX<sub>ox</sub> and (B) BOX<sub>ox</sub> in the spectral region from 1450 to 1560 cm<sup>-1</sup>, together with the component Lorentz bands obtained from band fitting.

	beef heart		Sulfolobus acidocaldarius	
	frequency	width	frequency	width
heme a, oxidized				
C=O (6cLS)	1649.0	9.2	1656	8
ν <sub>3</sub> (6cLS)	1499.1	14.8	1498	10
ν <sub>28</sub> (6cLS)	1470.1	13.1	1473	15
heme a <sub>3</sub> , oxidized				
C=O (6cHS)	1673.6	13.2	1673	7
C=O (5cHS)			1666	12
ν <sub>3</sub> (5cHS)			1490	11
ν <sub>3</sub> (6cHS)	1479.6	10.8	1482	10
heme a, reduced				
C=O (6cLS)	1609.2	12.6	1657	10
ν <sub>3</sub> (6cLS)	1490.8	6.9	1494	13
ν <sub>28</sub> (6cLS)	1466.1	16.0	1462	17
heme a <sub>3</sub> , reduced				
C=O (HS)	1664.7	9.4	1668	(9)
C=O (5cHS)			(1672)	(Ì0)
$\nu_3$ (HS)	1473.6	6.9	1476	11
ν <sub>3</sub> (5cHS)			(1469)	(13)

population. In addition, the intensity increase between 1620 and 1640 cm<sup>-1</sup> indicates the population of the 5cHS configuration of heme  $a_3$  ( $\nu_{10}$ ).

A reliable band fitting of the RR spectrum of  $SOX_{ox}$  is also possible above  $\sim 1630~cm^{-1}$ . This region, which includes the stretching vibrations of the formyl substituents, is shown in Figure 3 for both  $SOX_{ox}$  and  $BOX_{ox}$ . The spectral region above  $1650~cm^{-1}$  shows the presence of three bands in  $SOX_{ox}$ . The spectrum cannot be satisfactorily fitted without including the highest frequency band at  $1673~cm^{-1}$ . We assign all three of these bands as formyl stretching vibrations (Table I). The

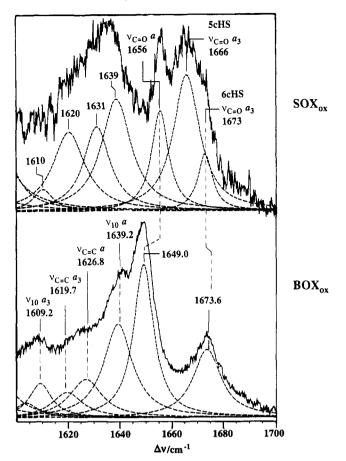


FIGURE 3: Resonance Raman spectra of (A) SOX<sub>ox</sub> and (B) BOX<sub>ox</sub> in the spectral region from 1600 to 1700 cm<sup>-1</sup>, together with the component Lorentz bands obtained from band fitting.

behavior of porphyrin a formyl stretching vibrations has been the object of a great deal of experimental effort (Babcock, 1988; Han et al., 1991). The heme a formyl stretch in BOX<sub>ox</sub> is located at 1649 cm<sup>-1</sup>, approximately 6 cm<sup>-1</sup> lower than that of a 6cLS model compound in aqueous medium (Han et al., 1991; Babcock, 1988). This downshift is attributed to strong hydrogen bonding within the protein. We assign the sharp peak at 1655 cm<sup>-1</sup> of SOX<sub>ox</sub> to a heme a formyl stretching vibration which is notably less hydrogen bonded than that of BOX<sub>ox</sub>, similar to an aqueous environment.

On the other hand, heme  $a_3$  of  $BOX_{ox}$  displays a formyl vibration at 1674 cm<sup>-1</sup>. It has yet to be shown that the formyl vibration can be used to predict the coordination state in porphyrin a model compounds. Thus, although the 5cHS porphyrin  $a^{3+}$  Cl<sup>-</sup>/CH<sub>2</sub>Cl<sub>2</sub> and the 6cHS porphyrin  $a^{3+}$ (DMSO)<sub>2</sub>/DMSO show formyl stretches at 1676 and 1672 cm-1, respectively (Callahan & Babcock, 1981), the different dielectric properties of methylene chloride and dimethyl sulfoxide may be responsible for this frequency shift, analogous to the work of Koyama et al. (1986) on metallochlorins. We are therefore unable to make a conclusive assignment for the bands at 1666 and 1673 cm<sup>-1</sup> with respect to the two coordination states of heme  $a_3$ , but it may be expected that the 5cHS heme  $a_3$  will show a formyl stretching vibration distinct from that of the 6cHS configuration. It seems reasonable to assume that the 1673-cm<sup>-1</sup> band, being closer in frequency to the  $BOX_{ox}$  heme  $a_3$  formyl stretch, corresponds to a formyl in a similar position as in standard 6cHS heme a<sub>3</sub> configuration found in BOX<sub>ox</sub> (Heibel et al., 1993; Babcock, 1988) as well as the quinol oxidase from Bacillus subtilis (Lauraeus et al., 1992). Then the band at 1666 cm<sup>-1</sup> may

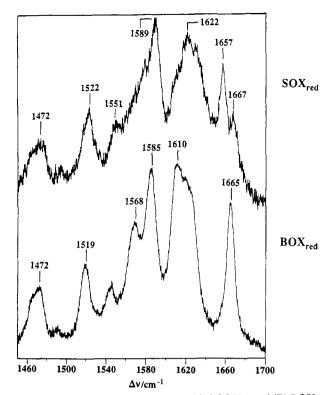


FIGURE 4: Resonance Raman spectra of (A) SOX<sub>red</sub> and (B) BOX<sub>red</sub> in the spectral region from 1450 to 1700 cm<sup>-1</sup> obtained with 413-nm

represent a formyl on a 5cHS heme a<sub>3</sub>. However, the relative intensity of RR bands cannot be regarded as a reliable guide to their relative populations.

The vinyl stretching region (1610–1640 cm<sup>-1</sup>) of SOX<sub>ox</sub> is clearly broader, relatively more intense, and less structured than in BOX<sub>ox</sub>. This can be ascribed to the additional contributions from the 5cHS species, so that the number of bands expected in this region is six instead of four in BOXox  $(\nu_{10} \text{ and } \nu_{C} = C \text{ for each species})$ . This spectral region has been fitted with the minimum number of components required to satisfactorily simulate the spectrum, implying that such components do not reflect real bands in each case. While the components at ~1610 and 1640 cm<sup>-1</sup> agree quite well with  $\nu_{10}$  bands of heme  $a_3$  and heme a of BOX<sub>ox</sub>, respectively, the components at ~1620 and 1630 cm<sup>-1</sup> include all three vinyl stretching vibrations and the  $\nu_{10}$  mode of the 5cHS heme  $a_3$ .

The Fully Reduced State. The RR spectrum of SOX<sub>red</sub> is shown in the marker band region (1440-1700 cm<sup>-1</sup>) in Figure 4A, again in comparison with the spectrum of the fully reduced beef heart oxidase, which has been analyzed in detail elsewhere (Heibel et al., 1993). In the mammalian enzyme, the spin configuration is preserved upon reduction for both hemes, although, unfortunately, the marker band frequencies of 6cHS and 5cHS ferrous porphyrins are very similar, so that a distinction between heme  $a_3$  coordination states solely from RR spectra is not possible. Mössbauer spectroscopy (Kent et al., 1982; Wang et al., 1988) and X-ray absorption (Scott, 1989) have revealed that heme  $a_3$  in BOX<sub>red</sub> has similar spectral characteristics as deoxymyoglobin, possessing a 5cHS configuration and being able to bind  $O_2$ .

The RR spectrum of SOX<sub>red</sub> displayed in Figure 4A shows some differences compared to that reported previously (Hildebrandt et al., 1991). The intensity of the sharp peak at 1657 cm<sup>-1</sup> as well as the intensity of the  $\nu_2$  envelope at 1589 cm<sup>-1</sup> relative to the  $\nu_{10}$ -vinyl stretching region (1610–1640 cm<sup>-1</sup>) is much higher in the present spectrum.

FIGURE 5: Resonance Raman spectra of (A) SOX<sub>red</sub> and (B) BOX<sub>red</sub> in the spectral region from 1450 to 1540 cm<sup>-1</sup>, together with the component Lorentz bands obtained from band fitting.

The inspection of BOX<sub>red</sub> and SOX<sub>red</sub> reveals obvious differences (Figure 4). Most of the spectral features are broader in SOX<sub>red</sub>, and several peaks are significantly shifted. The peak at 1568 cm<sup>-1</sup>, which in BOX<sub>red</sub> constituted an intense doublet with the 1588-cm<sup>-1</sup> peak, has lost intensity and is hidden as a broad and poorly resolved shoulder. Since in BOX<sub>red</sub> the main contribution to the 1570-cm<sup>-1</sup> peak comes from the  $v_2$  mode of heme  $a_3$ , we conclude that in SOX<sub>red</sub> the spectral differences are associated with the HS heme  $a_3$ . The lowered intensity of  $\sim 1570$  cm<sup>-1</sup> cannot be attributed to more unfavorable resonance conditions of heme  $a_3$ , since other modes, such as  $\nu_3$ , are not weaker than in BOX<sub>red</sub> (see below). Instead, these observations are reminiscent of the fully oxidized state where the broadening and intensity decrease of the 1570-cm<sup>-1</sup> peak has been ascribed to the overlapping contributions from the 5cHS and 6cHS configurations. Thus the question arises if, in analogy to the oxidized state, there is also an equilibrium between two HS forms in SOX<sub>red</sub>.

In fact, the analysis of the  $\nu_3$  band region which is presented in Figure 5A in an expanded view may support this idea. While the  $\nu_3$  mode of the 6cLS heme a is expected at  $\sim 1490$  cm<sup>-1</sup> and can readily be assigned to the weak band at 1494 cm<sup>-1</sup>, the corresponding mode of a 5cHS configuration is expected between 1470 and 1475 cm<sup>-1</sup>. In this part of the spectrum, the asymmetric peak at  $\sim 1470$  cm<sup>-1</sup> is similarly shaped but much broader than in BOX<sub>red</sub>. In the latter oxidase, this peak includes the mode  $\nu_{28}$  (heme a) and  $\nu_{3}$  (heme  $a_{3}$ ) at  $\sim 1466$  and 1474 cm<sup>-1</sup>, respectively (Table I). In SOX<sub>red</sub> a satisfactory fit is also achieved with two components which are, however, much broader than in BOX<sub>red</sub>. Thus it is possible that there is a third component which would correspond to a second HS form of heme  $a_3$ . The formyl and vinyl stretching

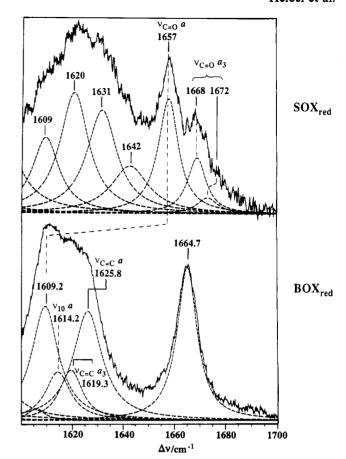


FIGURE 6: Resonance Raman spectra of (A) SOX<sub>red</sub> and (B) BOX<sub>red</sub> in the spectral region from 1600 to 1700 cm<sup>-1</sup>, together with the component Lorentz bands obtained from band fitting.

regions (1600–1700 cm<sup>-1</sup>) for reduced BOX<sub>red</sub> and SOX<sub>red</sub> are shown in Figure 6, revealing the most notable differences between the mammalian and Sulfolobus oxidases. The BOX<sub>red</sub> heme a formyl stretching vibration is assigned to a band at ca. 1610 cm<sup>-1</sup> (Babcock, 1988). We observe lower intensity in this region in the SOX<sub>red</sub> spectrum. We previously demonstrated that intensity at  $\sim 1610$  cm<sup>-1</sup> in SOX<sub>red</sub> is attributable to a heme  $a_3$  vibration ( $\nu_{10}$ ), because of lowered intensity upon CO binding in the reduced state (Hildebrandt et al., 1991). Thus, we conclude that, of the bands observed above 1650 cm<sup>-1</sup> in SOX, one should be assigned to the formyl stretching of heme a. This difference is a drastic departure from the classical cytochrome oxidase architecture (Babcock, 1988).

The peak at  $\sim 1670~\rm cm^{-1}$  is within the region expected for the formyl stretching of heme  $a_3$ , with the carbonyl group being in an non-hydrogen-bonding environment. Consequently, the remaining band in this region at  $1657~\rm cm^{-1}$  is ascribed to the formyl stretch of heme a. This frequency is about  $12~\rm cm^{-1}$  higher than those of 6cLS model compounds measured in nonaqueous media [1645 cm<sup>-1</sup> in CH<sub>2</sub>Cl<sub>2</sub> (Babcock, 1988); 1644 cm<sup>-1</sup> in DMSO (Choi et al., 1983)]. This implies that a non-hydrogen-bonding environment is not the only origin for this unusually high frequency. In addition, it may be that geometric distortion, i.e., an out-of-plane tilting of this substituent, weakens the coupling with the  $\pi$ -electron system of the porphyrin, bringing the carbonyl stretching frequency closer to that of an isolated aldehyde group (Dollish et al., 1974).

Also the band fitting analysis of the  $1670\text{-cm}^{-1}$  envelope does not provide an unambiguous conclusion whether or not there are two HS forms in the reduced heme  $a_3$ . As in the

 $\nu_3$  band region, the peak is broad and asymmetric but a fit with only one component gave similarly good standard deviations as the two-component fit which is shown in Figure

As emphasized for the oxidized state, the possible presence of a new coordination state for heme  $a_3$  renders the band fitting of the 1600-1650-cm<sup>-1</sup> spectral region of Figure 6 ambiguous. The band fitting of the vinyl stretching region in SOX<sub>red</sub> represents a minimum number of bands required to fit the spectrum. We note the extreme width of these bands and also that the spectral intensity at ca. 1642 cm1-1 is highly unusual, since heme skeletal vibrations are not found above 1640 cm<sup>-1</sup>. This band is probably due to a vinyl stretching

Structural Differences Compared to Other Oxidases. In the oxidized Sulfolobus enzyme, the frequency upshift of the formyl stretching mode of heme a compared to the bovine oxidase implies weaker hydrogen-bonding interactions of this substituent. This finding is in line with the blue-shift of the  $\alpha$ -absorption band by 4 nm to 597 nm (Babcock & Callahan, 1983). The Sulfolobus enzyme also differs from another quinol oxidase from B. subtilis which reveals a formyl stretching frequency similar to that of BOX<sub>ox</sub> (Lauraeus et al., 1992). On the other hand, a similarly high frequency as in SOX<sub>ox</sub> was also observed for the formyl stretching in wheat germ cytochrome oxidase (de Paula et al., 1990) which, however, shifts down to 1610 cm<sup>-1</sup> upon reduction as observed for all oxidases studied so far, including the quinol oxidase from B. subtilis (Babcock, 1988; Lauraeus et al., 1992). This is in sharp contrast to SOX<sub>red</sub>, where the stretching mode remains at the same position upon reduction. This implies that the redox-linked change of the hydrogen-bonding interactions of this substituent is not an unique structural property of the oxidase superfamily.

On the other hand, the present data do not provide evidence for significant structural differences of the porphyrin skeleton of heme a itself. The 1589-cm<sup>-1</sup> peak of  $SOX_{ox}$  is at a slightly higher frequency than the corresponding peak of BOX<sub>ox</sub> (Figure 1), which may be due to an upshift of the mode  $\nu_2$  of heme a. This as well as the upshift of  $\nu_{11}$  (Figure 2) may reflect differences in the electronic coupling between the porphyrin and the conjugated substituents (Choi et al., 1983; Willems & Bocian, 1984; Heibel et al., 1993). Presumably, these differences are not related to the unique hydrogenbonding interactions of the formyl group since a similarly high frequency of the  $v_2$  peak was also noted for quinol oxidase from B. subtilis (Laureaus et al., 1992).

The coexistence of 5cHS and 6cHS forms of oxidized heme  $a_3$  is the most pronounced difference compared to heme  $a_3$  of other cytochrome c oxidases as well as of the quinol oxidases from B. subtilis. This finding suggests that the interaction with the sixth ligand is weaker in SOX than in BOX. In this context, we wish to mention the unusual EPR spectroscopic behavior of heme a<sub>3</sub> of SOX<sub>ox</sub> (Anemüller & Schäfer, 1990; Anemüller et al., 1992). A strong signal at g = 6 was found which in BOX is only observed after breaking the antiferromagnetic coupling between heme a<sub>3</sub> and Cu<sub>B</sub> by partial reduction. A partial reduction of the heme groups in SOX<sub>ox</sub>, however, can be ruled out by the RR spectra. In particular, the v4-band region does not reveal a peak or a shoulder at ~1360 cm<sup>-1</sup>. A different EPR response was observed upon azide binding. A low spin signal at  $g_z = 2.9$  as noted in BOX<sub>ox</sub> was not found for the Sulfolobus enzyme which instead reveals a broadening of the g = 6 signal. Thus, these EPR data point to some structural anomalies of the heme a<sub>3</sub>-Cu<sub>B</sub> binuclear

center which may be related to the distortion of the coordination sphere of the heme  $a_3$  as revealed by the RR spectra. However, it should be noted that the distribution between the 5cHS and 6cHS forms is not necessarily the same in the EPR and RR experiments since both configurations may constitute a temperature-dependent equilibrium.

Both HS species exhibit different formyl stretching frequencies, which either reflects an intrinsic dependence of this mode on the coordination state (in analogy to the porphyrin marker bands) or, more likely, subtle differences in the interactions with the protein environment. This latter interpretation implies that both HS species differ not only in the proximity of the sixth ligand with respect to the heme iron but also in molecular structure in the vicinity of the formyl substituent. Following this idea, the two HS forms would reflect conformers rather than a simple coordination equilibrium. This would be the most plausible explanation if there were two heme a<sub>3</sub> HS species in the fully reduced state as

Finally, we wish to emphasize that the coexistence of two HS forms does not result from a partial denaturation of the enzyme during isolation and purification. It has been demonstrated that the Sulfolobus oxidase is constitutively active, indicating its functional and structural integrity (Gleissner et al., 1992).

The striking differences of the heme structures between SOX and BOX cannot be attributed to the different physiological reductants, i.e., caldariella quinone and cytochrome c, respectively. The far reaching similarities of the quinol oxidase from B. subtilis and the mammalian enzyme indicate that the specificity on the binding site does not affect the structure of the heme pocket. The same arguments also rule out that the lack of CuA in both quinol oxidases influences the active sites (Laureaus et al., 1992). It may be that the specific heme pocket structures in SOX may reflect the structural adaption of this enzyme to the extreme living conditions of this archaebacterium.

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